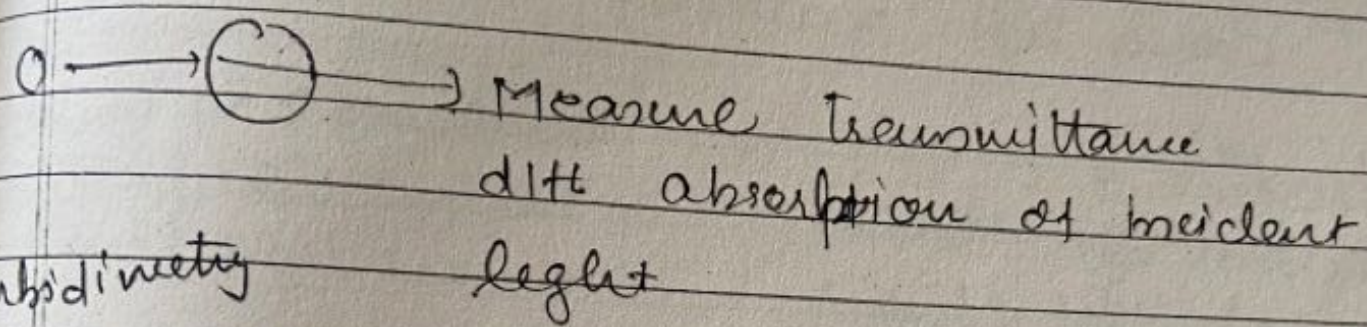
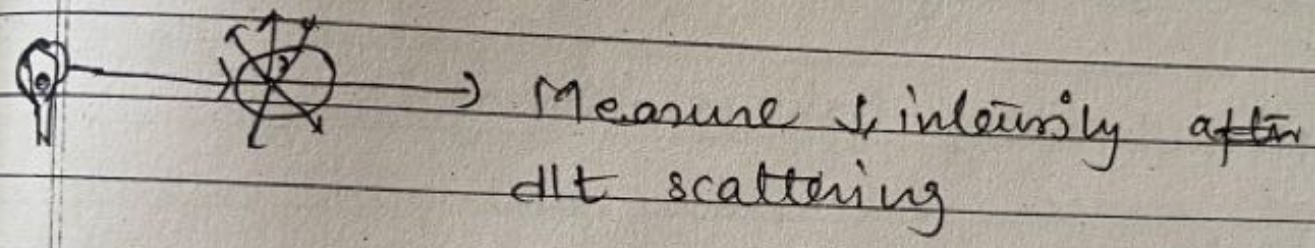


Nephelometry | Turbidimetry:

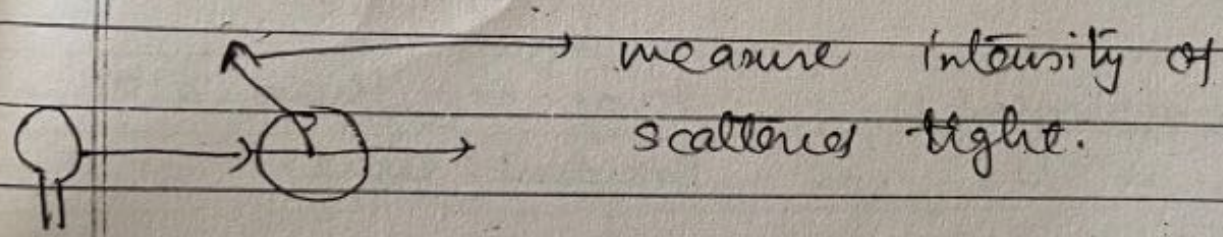
Absorbance.



turbidimetry



Nephelometry



Turbidimetry and Nephelometry.

21/01/06

Remember
principle

The radiant light when passes through a solution encounters a molecule in an elastic collision.

↓
causes light scattering in all direction.

→ fluorescent scattering	→	scattering
Incident light → low λ		Incident light → ^{same λ} low λ
emitted light → high λ		emitted light → high λ

→ Factors influencing light scattering.

① Particle size :-

Small size particle

↓
smaller than λ of incident light ($\frac{1}{10}$ th of λ)

↓
each particle is subjected to same electric field

↓
Reradiated and scattered light waves are in phase and reinforce each other.

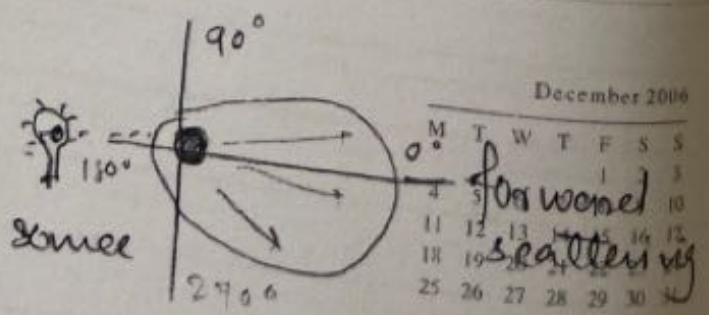
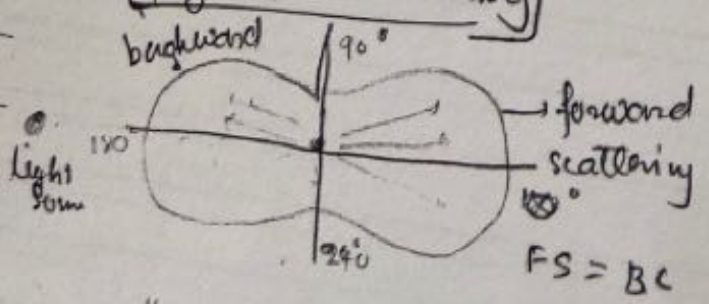
larger size particle

↓
large size than λ of incident light

↓
radiated wave no longer in phase

+
reinforcement in one direction
destructive interference in other direction.

Notes
Rayleigh scattering



FS > BS.

⇒ Theory of electrophoresis

Depending on charge they carry, ionised ^{solute} move towards cathode / anode in electrophoresis system.

ex → Positive ion → migrate to cathode.

Neg. ion → Anode.

• Ampholyte (protein, NA) → having either positive or negative charge.

↓
depends upon pH of solution

Acidic solutⁿ

↓

pH less than

its ~~pI~~ pI

↓

becomes +ve charged
migrate towards cathode

Alkaline solutⁿ

↓

pH more than its pI

↓

becomes -ve charged

↓

Anode.

→ Rate of migration:-

$$\text{driving force } F = xQ = \frac{(EMF)Q}{d}$$

where F = force exerted on ion.

x = current

Q = net charge

EMF = DC electro motive force (voltage)

d = length of electrophoresis medium

Basic Concept:-

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• Electrophoresis

migration of charged ~~mole~~ solutes / particles of any size in a liquid medium \downarrow influence of an electric field.

• Ionophoresis / isotachopheresis \rightarrow similar term but refers to migration of small ions.

• Zone electrophoresis

migration of charge protein molecules in porous supporting media like agarose gel film, such that each protein zone is sharply separated by zone of protein free area.

\downarrow
zones are visualized by staining \bar{c} protein specific stain

\downarrow
electrophoresis

\downarrow
scanned and quantified by densitometer

Use \rightarrow to separate protein in serum, urine, CSF, other ~~the~~ physiological fluids, RBC. tissue and uric acid.

→ steady acceleration of migration is counteracted by resisting force;

$$F' = 6\pi r \eta v$$

where F' = resisting force

r = ionic radius of solute

η = viscosity of medium.

v = rate of migration = length traveled / $\frac{\text{sec.}}{\text{cm}}$

$$F = F' \quad \text{b}$$

∴ resultant of 2 forces is constant velocity.

$$\therefore 6\pi r \eta v = x Q$$

$$\therefore \frac{v}{x} = \frac{Q}{6\pi r \eta} = \boxed{\mu}$$

→ μ is defined as electrophoretic mobility.

↳ rate of migration / unit ^{field} strength

directly
 $\mu \propto$ net charge.

inversely \propto to size of molecule and
viscosity of medium.

⇒ Factors affecting rate of mobility

- ① net charge of molecule
- ② size and shape of molecule
- ③ strength of electric field
- ④ support medium properties
- ⑤ ionic strength of buffer
- ⑥ Temp.

⑦ end osmosis

2 Electroendosmosis :-

electrophoretic support medium such as gel

↓
In contact with water, takes negative charge caused by adsorption of hydroxyl ion.

↓
Fixed to surface and immobile

↓
Positively ions in solution

* Wick flow - movement of support in to support medium ^{buffer}

Passage of current through resistive medium

↓
Heat generated

↓
evaporation of solvent from medium (dry)

↓
Draws buffer into medium

↓
Affect protein migration