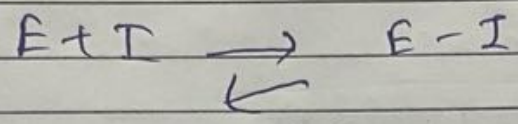
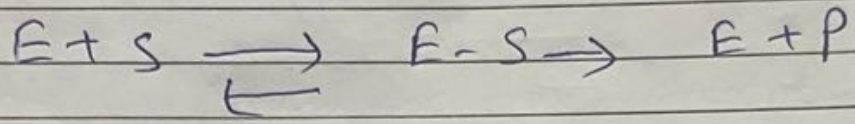


16) K_m and V_{max} in competitive and non-competitive enzyme inhibition.

→ Competitive Inhibition

- Inhibitor molecules are competing with the normal substrate molecule for binding to the active site of the enzyme, because the inhibitor is a structural analogue of the substrate.



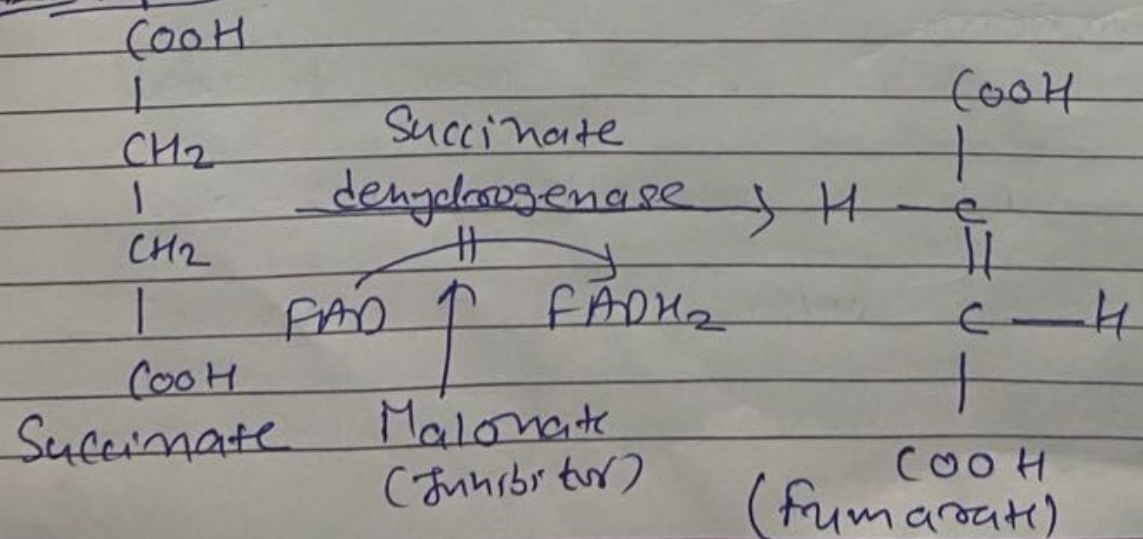
→ E-I (enzyme-inhibitor complex) can react only to reform the enzyme and inhibitor.

→ The number of enzyme molecules available for E-S formation is reduced.

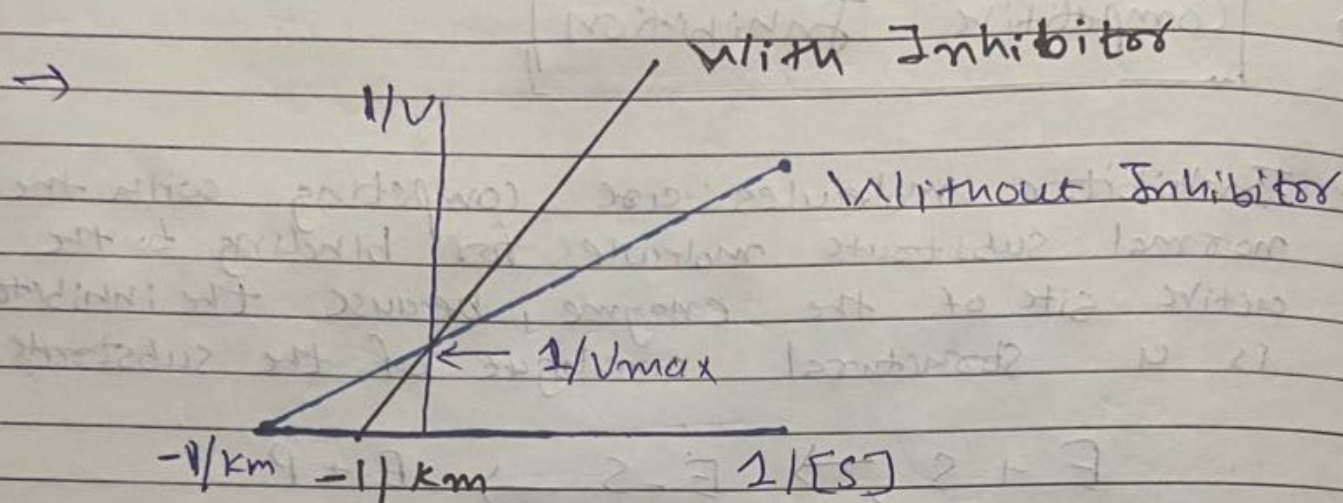
→ Since effective concentration of enzyme is reduced, the reaction velocity is decreased.

→ Similarity in 3 dimensional structure between substrate & inhibitor.

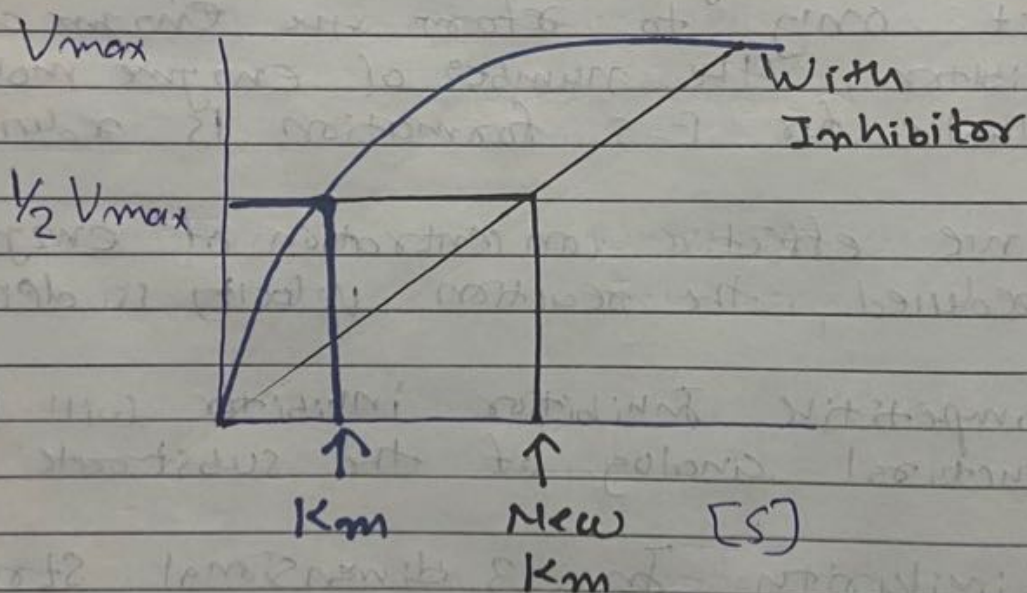
Example



→ Competitive inhibition is usually reversible or excess substrate abolishes the inhibition



(A) Lineweaver-Burk Plot



(B) Michaelis-Menten Saturation Curve

- From the above graph in competitive inhibition K_m is increased in presence of competitive inhibitor

- Affinity of the enzyme towards substrate is apparently decreased in presence of Inhibitor.

→ But V_{max} is not changed.

Non Competitive Inhibition (Irreversible)

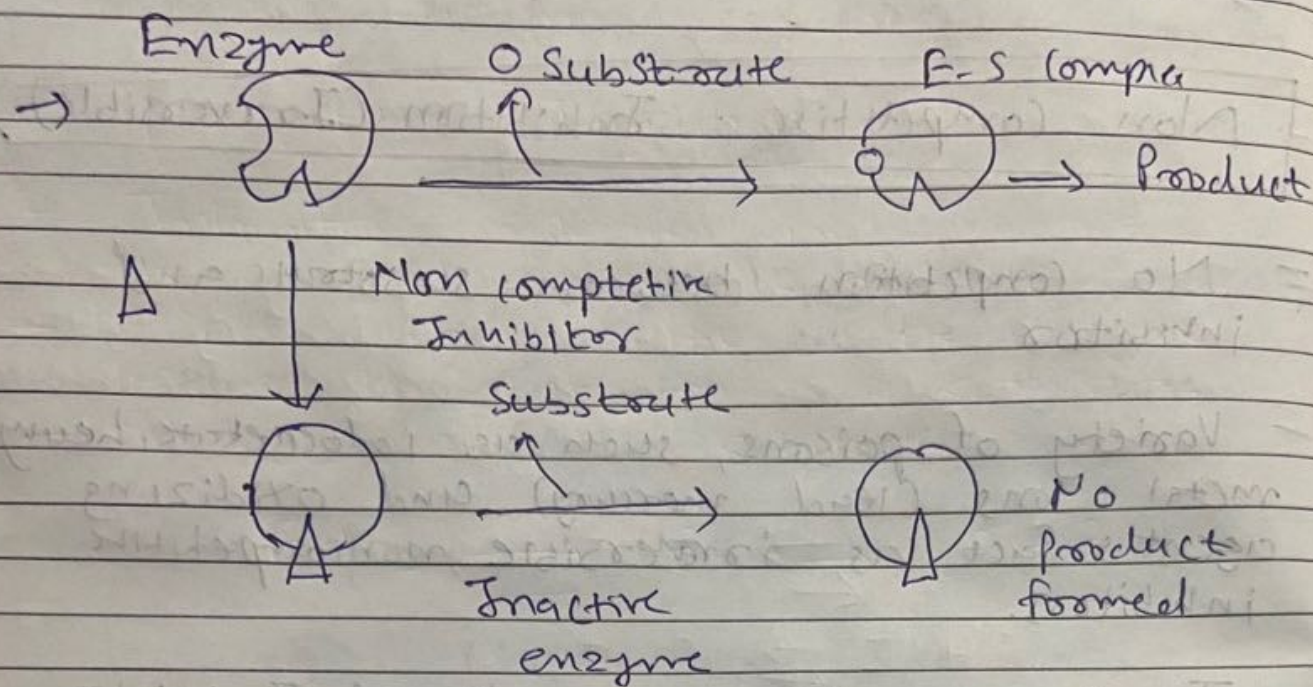
- No competition between substrate and inhibitor
- Variety of poisons, such as iodoacetate, heavy metal ions (lead mercury) and oxidizing agents act as irreversible noncompetitive inhibitors.
- It is also known as Mixed Inhibition.
- Inhibitor bind to different domain on the enzyme, other than the substrate binding site

- No structural resemblance to the substrate
an increase in the substrate concentration generally does not relieve the inhibition

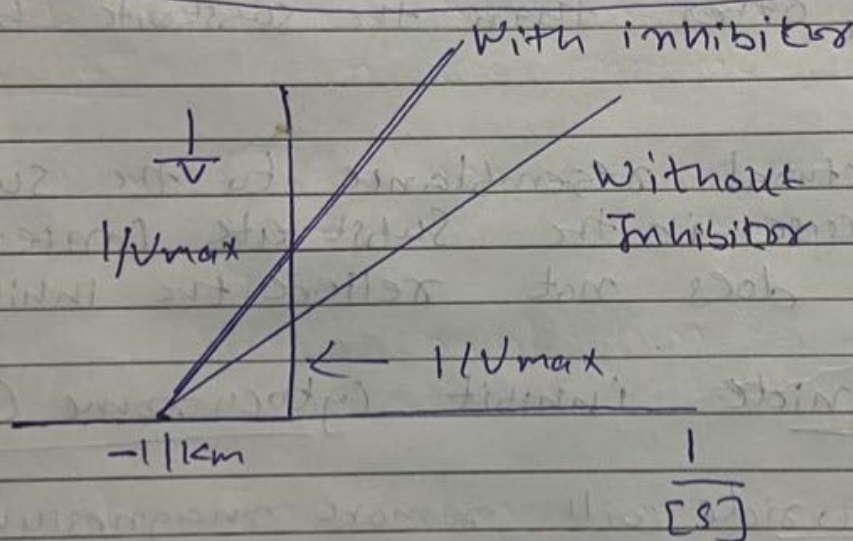
Ex - Cyanide inhibit Cytochrome Oxidase

- fluoride will remove magnesium and manganese ions and will inhibit the enzyme enolase and consequently the glycolysis.
- Iodoacetate would inhibit enzymes having -SH group in their active centers.

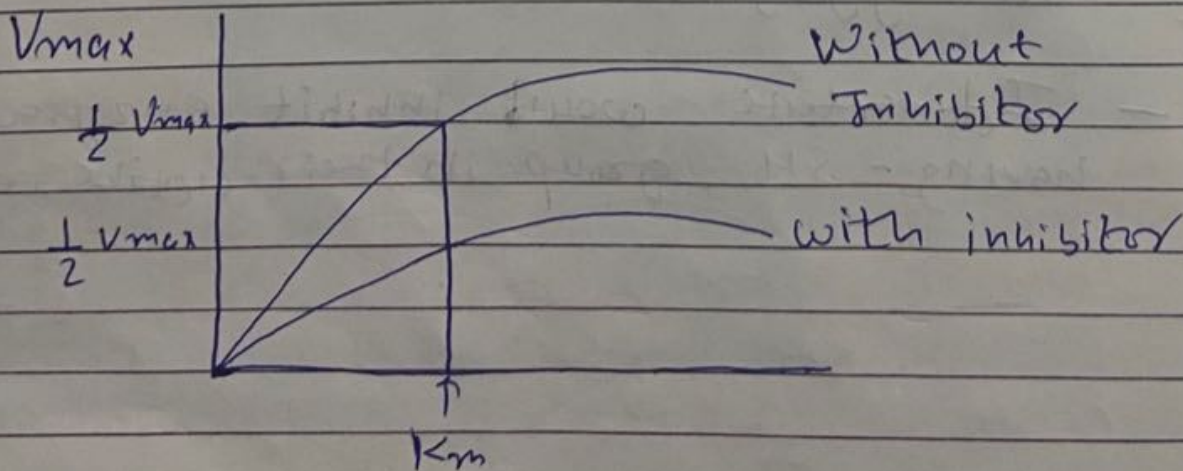
But V_{max} is not changed



(A) Lineweaver-Burst Plot



(b) Michaelis-Menten saturation curve



→ ~~The inhibitor~~

→ The inhibitor combines with the enzymes by forming a covalent bond and then the reaction becomes irreversible

→ The Velocity (V_{max}) is reduced, But K_m value is not changed.



The number of enzyme molecules available for the reaction is reduced

The number of enzyme molecules available for the reaction is reduced

The number of enzyme molecules available for the reaction is reduced

The number of enzyme molecules available for the reaction is reduced

The number of enzyme molecules available for the reaction is reduced

The number of enzyme molecules available for the reaction is reduced