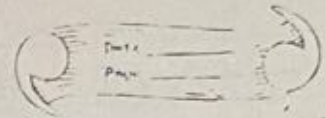


# Total protein



## ① Biouret method:-

→ Biouret is a same compound & give same colours. (Biouret also chelates  $Cu$ )

→ In strong alkalin cond<sup>n</sup> →  $Cu^+$  binds to 4 peptide bond

↓  
changes on absorp<sup>n</sup> from blue to purple.  
powderes ↳  $Cu$  ~~650~~ 540 nm.

→ Biouret<sup>n</sup> forms from Urea.

↓  
dry powderes heated into  $160^\circ C$  form biouret

→ Intestefence By some compound such as caseine, caseparagin, } due to binding to  $Cu^{+2}$   
TBS

→  $Cu^{+2}$  does n't react to A.A & dipeptide.

↓  
But some A.A who has min. to 2.  $NH_2$  group. →  $Cu^{+2}$  reacts.

→ peptide ~~with~~ which have proline.

↓  
Bind less to  $Cu^{+2}$

→ some side chain can contribute with binding to  $Cu^{+2}$

→ diff. protein has same reaction.

↓  
Only high conc. of some A.A interestese



→ More useful for purified protein.

→ Complex mixture → ~~Accuracy~~ & not good method.

→ At 220 to 245 nm

↓  
mainly peptide bond.

↓  
10 to 20 times greater sensitivity.

→ low mol. wt protein should be excluded  
1st to ↓ interference.

⑧ Kjeldahl method :-

↓  
ret. method.

→ Protein  $\xrightarrow{H_2SO_4}$  Ammonia

↓  
By Nessler's reagent.  
(like Besthelite)

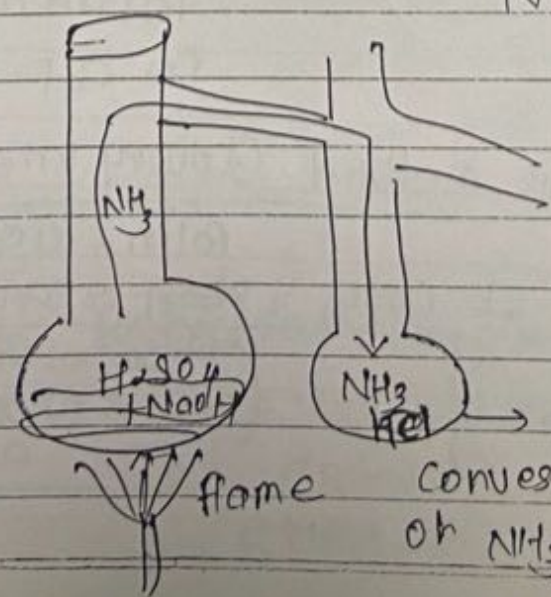
$NH_4^+$  Alkanilation  $\rightarrow NH_3$

↓  
Heat

↓  
evaporation

↓  
dissolved

↓  
acid titration



conversion  
of  $NH_3 \rightarrow NH_4^+$   
↓  
acid titration

↓  
When  $\text{NH}_3$  added into HCl

↓  
colours will change

↓  
Then titrate by using HCl  
(to take that original colour)

↓  
How much HCl is required  
is noted.

→ TP is not Homogeneous

↓ No SI unit

(Because it is compound) Not any std. standard.

④ Dye Binding method :-

Based on protein error (due to Buffering Capacity of protein)

~~cat~~  
Dye : CBB → Immunoglobulin gives 60%  
pyrogallol red → one of m. common  
method

↓  
to measure protein  
in CSF & Urine.

⑤ Folin-Ciocalteu method (Lowry method) :-

Serum  
(Peptides & protein  
containing tyrosine +  
tryptophan)

+  $\text{CuSO}_4$  + ~~phos~~ reagent ——— alkaline  
medium  
(phosphotungstic  
acid & phosphomolybdic  
acid)

Cu<sup>2+</sup> - peptide contain  
Tyrosin & tryptophan  
complex

Tungsten Blue +  
molybdate Blue

measures at 650 to 750  
nm.

forms due to reduction  
reaction.

→ Interference :- due to LMW compounds  
like tryptophan & Tyrosin  
→ salicylate } drugs  
sulfa }  
→ phenolic compounds

→ IF urine containing high compound on  
phenol ↓

removal of phenolic compound should  
do before protein estimation.

⑥ Refractometry :

↓  
rapid estimate of high conc. of TP.  
( > 35 g/L )

if conc. ↓ < 35 g/L → accuracy ↓

→ Refractometry is mainly used in lab

↓  
To determine conc. of solute in  
~~urine~~ urine

⑦ Turbidimetric & Nephelometric method:

↓  
protein are aggregated by using various agents like TCA, sulfosacrylic acid

↓  
form<sup>n</sup> of suspension of uniform Insoluble protein particles

↓  
Scattering of incident light.

Weber  
CGF & others

\* Reference Interval :-

↓  
5.5 to 7.5 g/dL

→ plasma contain 0.3 g/dl higher protein than serum

↓  
plasma contain fibrinogen & clotting factors.

→ f Protein :- in hemocentration<sup>n</sup>  
in vomiting,  
diarrhoea,  
addison's d'se  
DKA

→ Hemodilution ? water intoxication  
Salt retention syndrome  
massive IV infusion  
decumbent position